

Lecture Time, Location Tuesday, Thursday 07:30 – 09:00 P.M.  
Elmendorf Education Building, Room

Instructor Thomas H. Morse

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Office Hours Tuesday, Thursday 07:00 – 07:30  
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Textbook UAA Custom Edition of the 4<sup>th</sup> Edition of  
Introductory Chemistry, by Nivaldo Tro

Material Covered Chapters 1 – 11, Selected Sections of Chaps. 12, 13  
(not tested on Chap. 12)

<u>Grading Policy</u>	3 Semester Exams	300 pts.
	1 Final Exam (Comprehensive)	100 pts.
	<u>Total</u>	<u>400 pts.</u>

Grades are curved via a normal Gaussian Distribution

Add/Drop Deadline Sep 06, 2013 (100% Refund, Removal from Transcript)

Withdrawal,  
Credit to Audit Deadlines Nov. 15, 2013

Final Exam Thursday Dec. 12, 2013, at regular class time.

"Nothing is to be feared, only understood." Marie Curie

"The supreme principle of education –  
to offer food only to those who hunger for it" Frederick Nietzsche

Course Objectives - To give students a thorough grounding in introductory inorganic chemistry (both theoretical and computational). This will prepare them for future chemistry courses, in particular the Chemistry 103-104 or Chemistry 105-106 sequences at UAA.

Semester Schedule for Fall, 2013 (Tentative, subject to change)

~~Aug. 27~~ Chapters 1, 2  
~~Aug. 29~~ Chapter 2  
~~Sep. 03~~ No Class  
~~Sep. 05~~ Chapters 2, 3  
~~Sep. 10~~ Chapter 3  
~~Sep. 12~~ Chapters 3, 4  
~~Sep. 17~~ Chapter 4  
~~Sep. 19~~ Chapter 4  
~~Sep. 24~~ Exam #1 (Chapters 1 - 3)  
~~Sep. 26~~ Chapter 5  
~~Oct. 01~~ Chapter 5  
~~Oct. 03~~ Chapter 6  
~~Oct. 08~~ Chapter 6  
~~Oct. 10~~ Chapter 7  
~~Oct. 15~~ Chapter 7  
~~Oct. 17~~ Exam #2 (Chapters 4 - 6)  
~~Oct. 22~~ Chapter 8  
~~Oct. 24~~ Chapter 8  
~~Oct. 29~~ Chapter 9  
~~Oct. 31~~ Chapter 9  
~~Nov. 05~~ Chapter 10  
~~Nov. 07~~ Chapter 10  
Nov. 12 Exam #3 (Chapters 7 - 9)  
Nov. 14 Chapter 10, Essential Fatty Acids Lecture  
Nov. 19 Chapter 11  
Nov. 21 Chapter 11  
Nov. 26 Chapter 11  
Nov. 28 No Class - Thanksgiving Holiday  
Dec. 03 Chapters 12, 13  
Dec. 05 Chapter 13, Nuclear Radiation Lecture  
Dec. 10 Open Review  
Dec. 12 Final Exam (Comprehensive)

## Chemistry 055 Final Exam Study Guide

- 1) Density Calculations – Chapter 2
- 2) Specific Heat Calculations – Chapter 3
- 3) Balancing Chemical Equations – Chapter 7
- 4) Mass-to-Mass Calculations – Chapter 8
- 5) Valance (Outer Shell) Electrons and Columns in the Periodic Table – Chapter 9
- 6) Use of Electron Filling Chart – Chapter 9
- 7) Ability to solve for any variable in the equation  $v=(f) \cdot (\lambda)$  – Chapter 9
- 8) Electronegativity and Covalent, Polar Covalent, Ionic Bonds – Chapter 10
- 9) Lewis Structures, Resonance Structures – Chapter 10
- 10) Gas Laws – Chapter 11
- 11) Pressure Unit Conversions, STP – Chapter 11
- 12) Law of Partial Pressures – Chapter 11
- 13) Percentage of Elements in Air – 78% Nitrogen ( $N_2$ ), 21% Oxygen ( $O_2$ )
- 14) Molarity Calculations, Molarity Dilutions – Chapter 13, Lecture Notes
- 15) Percent Mass Calculations – Lecture Notes
- 16) Ability to use Solubility Tables on P. 217 of Text (Chapter 7)